# **Origins of Solar Systems**

Lecture held for the International Max Planck Research School "Solar System and beyond" February 13-15, 2006

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including student lectures by Clementina Sasso Lotfi Yelles Chaouche Ingo von Borstel Stefan Schröder Esa Vilenius Emre Isik Silvia Protopapa Elias Roussos

# Origin of solar systems: Organization

# Lecture (KJ):

Introduction and overview Dense molecular clouds, photodissociation regions and protostars

Protoplanetary disks Equilibrium condensation of a solar nebula

Meteorites and the early solar system

Origin of giant planets

Comets and the early solar system

#### Student talks:

Origin of the elements and Standard Abundance Distribution

Agglomeration of planetesimals and protoplanets

Isotope chronology of meteorites and oxygen isotopes

Extrasolar planets

**Transneptunian Objects** 

#### Schedule:

Monday und Tuesday: 9:30-12:15 (3\*45 Min plus 2\*15 Min break) 14:00-16:45 Wednesday: 9:30-12:15

Wednesday afternoon: Seminar and Colloquium

#### Solar system formation Early stages (Ewine van Dieshoek, Leiden) a-d •A lot of steps between d and e: •Protoplanetary disk is hot near the Sun and cold far from the Sun, condensation of gas depending on temperature · Formation and growth of planetesimals (strongly 1 pr 10 000 AU dependent on relative velocity) • Formation of terrestrial and giant planets • Early Jupiter prevents planetesimal growth in its neighborhood $\rightarrow$ origin of asteroids Comets originate in the Kuiper belt at about 40 AU from the Sun •Long-period comets are scattered into Oort cloud, disturbed and isotropized by the influence of 100 AU 1 - 10 4-10 <sup>5</sup>year -1 t - 10 6-10 7 year passing stars and the galactic bulge. •Short-period comets go directly from Kuiper belt to H,O, CH,OH, the inner solar system H2CO, HCN, . •Meteorites come from the surfaces of asteroids and d from Mars to the Earth. Their measurement in the laboratory has contributed greatly to our knowledge 50 AU about solar system formation. 1>107 yes

Excurse on blackbody radiation. Radiation laws related to Planck's law  $B_{\nu}(T)d\nu = \frac{2h\nu^{3}}{c^{2}} \frac{1}{e^{h\nu/kT} - 1} d\nu \quad \text{W m}^{-2} \text{ rad}^{-2}$ or, using  $B_{\lambda}(T)d\lambda = B_{\nu}(T)d\lambda \cdot c/\lambda^{2}$   $B_{\lambda}(T)d\lambda = \frac{2hc^{2}}{\lambda^{5}} \frac{1}{e^{hc/\lambda kT} - 1} d\lambda \quad \text{W m}^{-2} \text{ rad}^{-2}$   $\frac{h\nu}{kT} \gg 1 \qquad B_{\nu}(T) \approx \frac{2h\nu^{3}}{c^{2}} e^{-h\nu/kT} d\nu \quad \text{Wien}$   $\frac{h\nu}{kT} \ll 1 \qquad B_{\nu}(T) \approx \frac{2\nu^{2}kT}{c^{2}} \quad \text{Rayleigh-Jeans}$ Wien's displacement law:  $\frac{c}{\nu_{max}} \cdot T = 5.10 \cdot 10^{-3} \quad [\text{meter K}] \quad \text{or}$   $\lambda_{max} \cdot T = 2.90 \cdot 10^{-3} \quad [\text{meter K}]$ With T = 290 K (room temperature) we get  $\lambda_{max} = 10^{-5} \text{ m} = 10 \,\mu\text{m}$ . With T = 5800 K (solar-type star) we get  $\lambda_{max} = 0.5 \,\mu\text{m}$ .





| Diffe                                 | Different phases of the interstellar medium   |                |                          |                         |              |                                  |                                    |  |
|---------------------------------------|---|----------------|--------------------------|-------------------------|--------------|----------------------------------|------------------------------------|--|
| Dillo                                 | ient price  |                |                          |                         |              |                                  |                                    |  |
|                                       | Physical Characteristics of Molecular Regions in the Interstellar Medium <sup>a</sup> |                |                          |                         |              |                                  |                                    |  |
|                                       | Density<br>(cm <sup>-3</sup> )  | Т<br>(К)       | Mass<br>M <sub>O</sub>   | A <sub>V</sub><br>(mag) | Size<br>(pc) | $\Delta V$ (km s <sup>-1</sup> ) | Examples                           |  |
| Diffuse Clouds                        | 100 - 800   | 30 - 80        | 1 - 100                  | ≲1                      | 1-5          | 0.5 - 3                          | ζ Oph                              |  |
| Translucent Clouds                    | 500 - 5000  | 15 - 50        | 3 - 100                  | 1 – 5                   | 0.5 - 5      | 0.5 – 3                          | HD 169454;<br>High-latitude clouds |  |
| Cold, Dark Clouds                     |   |                |                          |                         |              |                                  |                                    |  |
| complex                               | $10^2 - 10^3$   | ≳ 10           | $10^3 - 10^4$            | 1 - 2                   | 6 - 20       | 1 – 3                            | Taurus-Auriga                      |  |
| clouds                                | $10^2 - 10^4$   | ≳ 10           | $10 - 10^3$              | 2 - 5                   | 0.2 - 4      | 0.5 - 1.5                        | B1, B5                             |  |
| cores/clumps                          | $10^4 - 10^5$   | $\approx 10$   | 0.3 - 10                 | 5 - 25                  | 0.05 - 0.4   | 0.2 - 0.4                        | TMC-1, B335                        |  |
| Giant Molecular Clouds                |   |                |                          |                         |              |                                  |                                    |  |
| complex                               | 100 - 300   | 15 - 20        | $10^{5}-3 \times 10^{6}$ | 1 - 2                   | 20 - 80      | 6 - 15                           | M 17, Orion                        |  |
| clouds                                | $10^2 - 10^4$   | $\gtrsim 20$   | $10^3 - 10^5$            | $\gtrsim 2$             | 3 - 20       | 3 - 12                           | Orion OMC-1, W3 A                  |  |
| warm clumps                           | $10^4 - 10^7$   | 25 - 70        | $1 - 10^{3}$             | 5 - 1000                | 0.05 - 3     | 1 - 3                            | M 17 clumps, Orion 1/5 S           |  |
| hot cores                             | $10^7 - 10^9$   | 100 - 200      | $10 - 10^3$              | 50 - 1000               | 0.05 - 1     | 1 - 10                           | Orion hot core                     |  |
| <sup>a</sup> Table adapted from Golds | mith (1987), Tu   | mer (1989a) ar | d Friberg and Hja        | Imarson (1990).         |              |                                  |                                    |  |

Van Dishoek E.F. et al., in Levy and Lunine eds. "Protostars and Planets III", U. of Az. Press, 1993, pp 163-241.

# Van Dishoek E.F. et al., in Levy and Lunine eds. "Protostars and Planets III", U. of Az. Press, 1993.

#### TABLE I Angular Sizes of Protostellar Objects and Capabilities of Telescopes Linear size Angular Size Telescope<sup>b</sup> Angular Resolution Taurus Orion M17 Galactic Center 115 230 345 810 140 pc 450 pc 2.2 kpc 8.5 kpc GHz GHz GHz GHz 5 AU 0."04 0."01 45 m (Nobeyama) 15 Inner solar nebula 100 AU 0″7 0."2 0."05 30 m (IRAM) \_ 22" 12" 7" Outer solar nebula 1000 AU 7″ 2″ 0.5 0"1 15 m (JCMT/SEST) 44' 20″ 15″ 6″ Presolar nebula 0.05 pc Cloud core 5″ 74″ 23" 1″ 10 m (CSO) 30 20' 9' 0.5 pc 12' 4′ 50″ 12″ Interferometer (OVRO/ 4-7" 1–2″ Cloud BIMA/Nobeyama/IRAM <sup>a</sup> The table only lists the capabilities of currently operating telescopes, not those of future projects. <sup>b</sup> IRAM = Institute de Radio Astronomie Millimetrique; ICMT = James Clerk Maxwell Telescope; CSO = Caltech Submillimeter Observatory; SEST = Swedish-ESO Submillimeter Telescope; OVRO = Owens Valley Radio Observatory; BIMA = Berkeley-Illinois-Maryland Array.

The Taurus molecular cloud in CO emission.

Note the large extent of the cloud and the many small objects embedded in it.

The map has been done in the light of the CO molecule (millimeter wavelength).

CO is a very important interstellar molecule and believed to be a tracer of the (unobservable) H<sub>2</sub> molecule.



Figure 1. Velocity integrated intensity of CO emission in the Taurus molecular cloud complex. The lowest contour is  $0.5 \text{ K km s}^{-1}$ , and the separation between contours is  $1.5 \text{ K km s}^{-1}$ . The border of the surveyed region is indicated by the outer solid line. Various clouds such as B5 and cloud cores such as TMC-1 discussed in the text are indicated (figure adapted from Ungerechts and Thaddeus 1987).

| Туре   | Process  | Rate Coefficien                                |
|--|--|--|
|  | Formation Processes  |  |
| Radiative association  | $X + Y \rightarrow XY + h\nu$  | $10^{-16} - 10^{-9}$                           |
| Grain surface formation  | $X + Y: g \rightarrow XY + g$  | $\sim 10^{-18}$                                |
|  | Destruction Processes  |  |
| Photodissociation  | $XY + h\nu \rightarrow X + Y$  | $\sim 10^{-10} - 10^{-8}$ s                    |
| Dissociative recombination   | $XY^+ + e \rightarrow X + Y$   | $\sim 10^{-6}$                                 |
| Collisional dissociation   | $XY + M \rightarrow X + Y + M$   |  |
|  | Chemical Processes   | ,  |
| Ion-molecule exchange  | $X^+ + YZ \rightarrow XY^+ + Z$  | $\sim 10^{-9}$                                 |
| Charge-transfer  | $X^+ + YZ \rightarrow X + YZ^+$  | $\sim 10^{-9}$                                 |
| Neutral-neutral  | $X + YZ \rightarrow XY + Z$  | $\sim 10^{-12}$                                |
| <sup>a</sup> Approximate rate coefficients<br>ficients are sensitive to tempe<br>the unattenuated interstellar r | s appropriate for cold dark clc<br>rature. For photodissociation<br>adiation field are listed. | ouds. All rate coef-, the rates in $s^{-1}$ in |

#### Build-up of complex molecules in dense molecular clouds:

• In dense molecular clouds the only available energy to activate molecules comes from cosmic rays

- Cosmic rays penetrate into molecular clouds up to column densities of 10<sup>24</sup> cm<sup>-2</sup>.
- Cosmic rays ionize He and  $H_2$  at rates 10<sup>-17</sup> to 10<sup>-16</sup> s<sup>-1</sup>.
- $H_3^+$  is formed:  $H_2^+ + H_2^+ \rightarrow H_3^+ + H$  or  $H_2^- + He^+ \rightarrow H_3^+ + H + He$ .

 $\rm H_3^+$  is an interesting "floppy" ring molecule. It has transitions in the L and K bands. It was first discovered on Jupiter, and in the 90<sup>th</sup> was finally discovered in the interstellar medium.

H<sub>3</sub><sup>+</sup> is the key to molecule formation in dark clouds.





| Van Dishoek E.F. et al., | in Levy and Lunine eds. | "Protostars and | Planets III" |
|--------------------------|-------------------------|-----------------|--------------|
| U. of Az. Press. 1993.   |                         |                 |              |

| TABLE III       Identified Interstellar and Circumstellar Molecules <sup>a</sup> |                     |                               |                                    |                                    |                      |
|--|---------------------|-------------------------------|------------------------------------|------------------------------------|----------------------|
| Species  | Name                | Species                       | Name                               | Species                            | Name                 |
| H <sub>2</sub>   | molecular hydrogen  | C <sub>2</sub> H <sub>2</sub> | acetylene                          | C <sub>6</sub> H                   |                      |
| C <sub>2</sub>   | diatomic carbon     | C <sub>3</sub> H              | propynylidyne $(l \text{ and } c)$ | CH <sub>2</sub> CHCN               | vinyl cyanide        |
| CH   | methylidyne         | H <sub>2</sub> CO             | formaldehyde                       | CH <sub>3</sub> C <sub>2</sub> H   | methylacetylene      |
| CH <sup>+</sup>  | methylidyne ion     | NH <sub>3</sub>               | ammonia                            | CH <sub>3</sub> CHO                | acetaldehyde         |
| CN   | cyanogen            | HNCO                          | isocyanic acid                     | CH <sub>3</sub> NH <sub>2</sub>    | methylamine          |
| CO   | carbon monoxide     | HOCO <sup>+</sup>             | protonated carbon dioxide          | HC <sub>5</sub> N                  | cvanodiacetylene     |
| CS   | carbon monosulfide  | HCNH <sup>+</sup>             | protonated hydrogen cyanide        | .,                                 | .,                   |
| OH   | hydroxyl            | HNCS                          | isothiocyanic acid                 |                                    |                      |
| HCI  | hydrogen chloride   | C <sub>3</sub> N              | cvanoethynyl                       | HCOOCH <sub>2</sub>                | methyl formate       |
| NO   | nitric oxide        | C <sub>1</sub> O              | tricarbon monoxide                 | CH <sub>2</sub> C <sub>2</sub> N   | methylcyanoacetylene |
| NS   | nitrogen sulfide    | H <sub>2</sub> CS             | thioformaldehvde                   | CH <sub>2</sub> C <sub>4</sub> H   | methyldiacetylene    |
| SiC  | silicon carbide*    | H <sub>3</sub> O <sup>+</sup> | hydronium ion                      | CH <sub>3</sub> CH <sub>3</sub> O  | dimethyl ether       |
| SiO  | silicon monoxide    | C <sub>3</sub> S              | 2                                  | CH <sub>3</sub> CH <sub>2</sub> CN | ethyl cyanide        |
| SiS  | silicon sulfide     | HC <sub>2</sub> N             | •                                  | CH <sub>2</sub> CH <sub>2</sub> OH | ethanol              |
| SO   | sulfur monoxide     | -                             |                                    | HC <sub>7</sub> N                  | cyanobexatrivne      |
| PN   |                     |                               |                                    |                                    | ejjalonenalijne      |
| CP   | * 1                 | C₄H                           | butadivnyl                         |                                    |                      |
| so+  | sulfoxide ion       | C <sub>3</sub> H <sub>2</sub> | cvclopropenvlidene                 | CH <sub>2</sub> C <sub>4</sub> CN  |                      |
| NaCl   | sodium chloride*    | H <sub>2</sub> CCC            | propadienvlidene                   | CH <sub>2</sub> CH <sub>2</sub> CO | acetone              |
| AlCl   | aluminum chloride*  | HCOOH                         | formic acid                        | engengee                           | averene              |
| KCI  | potassium chloride* | CH <sub>2</sub> CO            | ketene                             |                                    |                      |
| AIF  | aluminum fluoride*† | HC <sub>3</sub> N             | cvanoacetylene                     | HC <sub>0</sub> N                  | cvano-octa-tetra-vne |
| NH .   | nitrogen hydride    | CH <sub>2</sub> CN            | cyanomethyl                        |                                    | eyano com-totta-ync  |

|                               |                            | . N                 | 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1 |                    |                     |
|-------------------------------|----------------------------|---------------------|--|--------------------|---------------------|
| SiN                           | •                          | NH <sub>2</sub> CN  | cyanamide                                |                    |                     |
|                               |                            | CH <sub>2</sub> NH  | methanimine                              | HC <sub>11</sub> N | cyano-deca-penta-yr |
| $H_2D^+$                      | <b>†</b>                   |                     |  | CH4                | methane             |
| C <sub>2</sub> H              | ethynyl                    |                     |  | SiH <sub>4</sub>   | silane*             |
| CH <sub>2</sub>               | methylene <sup>†</sup>     |                     |  | C <sub>4</sub> Si  | · •                 |
| HCN                           | hydrogen cyanide           |                     |  | C₅                 | pentatomic carbon*  |
| HNC                           | hydrogen isocyanide        | C₅H                 | pentynylidyne                            | HCCNC              | isocyanoacetylene   |
| HCO                           | formyl                     | $C_2H_4$            | ethylene*                                |                    |                     |
| HCO <sup>+</sup>              | formyl ion                 | H <sub>2</sub> CCCC | butatrienylidene                         |                    |                     |
| HOC <sup>+</sup>              | isoformyl ion <sup>†</sup> | CH <sub>3</sub> OH  | methanol                                 |                    |                     |
| N <sub>2</sub> H <sup>+</sup> | protonated nitrogen        | CH <sub>3</sub> CN  | methyl cyanide                           |                    |                     |
| HNO                           | nitroxyl                   | CH <sub>3</sub> NC  | methyl isocyanide                        |                    |                     |
| H <sub>2</sub> O              | water                      | CH <sub>3</sub> SH  | methyl mercaptan                         |                    |                     |
| HCS <sup>+</sup>              | thioformyl ion             | NH <sub>2</sub> CHO | formamide                                |                    |                     |
| H <sub>2</sub> S              | hydrogen sulfide           | HC <sub>3</sub> HO  | propynal                                 |                    |                     |
| OCS                           | carbonyl sulfide           |                     |  |                    |                     |
| SO <sub>2</sub>               | sulfur dioxide             |                     |  |                    |                     |
| SiC <sub>2</sub>              | silicon dicarbide*         |                     |  |                    |                     |
| $C_2O$                        | dicarbon monoxide          |                     |  |                    |                     |
| C3                            | triatomic carbon*          |                     |  |                    |                     |
| C <sub>2</sub> S              |                            |                     |  |                    |                     |

### Interstellar chemistry: Photodissociation regions

UV radiation of mass-rich, early-type stars evaporates grains and dissociates molecules.

The "elephant trunks" hide condensations of newly forming stars. These clouds are denser and therefore resist the evaporation and dissociation longer.











#### Cloud collapse:

Force balance: pressure and centrifugal force versus gravitation.

*Virial theorem:* Multiply the equation of motion with the radius vector. "Virial" = torque = Drehmoment see J. Lequeux "The interstellar medium", Springer 2003, Chap. 14.

In the absence of external pressure  $E_{Grav} = -2E_{Kin}$ 

Jeans mass  $M_J$ :  $M_J \approx \left(\frac{kT}{G\mu_a m_{amu}}\right)^{3/2} \frac{1}{\sqrt{\rho}}.$ 

The less dense the object is the more massive it must be to collapse. Examples: galaxies, star clusters, stars, planets. The cooler the object the easier it collapses.  $n > 10^{-11}$  g cm<sup>-3</sup> and T=10K needed to form a Jupiter size planet. This density is much larger than that observed in interstellar clouds .

Gravitational collapse may be triggered by supernova explosion or a galactic density wave.



#### Collapse of molecular cloud cores and star formation

Free fall time scale:

 $t_{ff} = \left(\frac{3\pi}{32G\rho}\right)^{1/2}$ 

For the Sun it is about 30 minutes.

Clumps are densest near their centers, collapse is inside-out. Angular momentum conversation leads to spin-up of the cloud, may cause the collapse to stop and to fragment the cloud. Therefore most often multiple stellar systems form.

Virtually all single stars and many multiple stars are surrounded by a flat disk during formation. The mass collects in the central star and the angular momentum in the disk.

In solar system 99.8% of the mass is in the central star and 98% of the angular momentum is in planetary orbits.

Temperature and density rise during collapse. Cloud becomes opaque, pressure builds up, D is burnt into He. When D is exhausted, star shrinks and heats further up, <sup>1</sup>H fusion starts.

Early phases of star formation can be observed in the microwave range, which presently is a very active part of research. Later the star becomes visible in the IR wavelength range.





Rotating protostellar disk DM Tau, obserded in millimeter range and presented here as three-dimensional spectrum. The system is well fitted by Keplerian rotation V~r<sup>-0.5±0.1</sup> Radius 525AU, Mass = 0.6-0.85 M<sub>p</sub> Mundy et al., in Mannings, et al. eds. "Protostars and Planets IV", U.of Az. Press, 2000, 355-376.



Figure 6. Images of the CO J = 2-1 emission, model, and residuals for the DM Tauri system from Guilloteau and Dutrey (1998). The upper set of panels shows the observed emission, with each panel labeled with velocity. The middle panels display the emission from the best-fitted model. The lower panels display the differences between the observations and the model. The angular resolution of the observations is  $3.5 \times 2.4''$ .

Lower panel not shown

## **Orion Nebula**

Note that HII region is *in front* of dark cloud!

see e.g. : Henbest N., Marten M., The New Astronomy, Cambridge 1983













Credit: Mark McCaughrean (Astrophysikalisches Institut Potsdam), Hans Zinnecker (Astrophysikalisches Institut Potsdam), and John Rayner (University of Hawaii)







#### Early evolution of a collapsing star (continued):

A star of 15 solar masses condenses to the main sequence in 60000 years, for a star with 0.1 solar masses the process takes hundreds of millions of years.

Very young stars (T Tauri stars) difficult to observe as they are enshrouded into dense clouds.

T Tauri stars have high Lithium content (central temperature not yet high enough to destroy Lithium).

#### Bonnor-Ebert-Sphere:

Equilibrium isothermal sphere of finite size with a fixed temperature and boundary pressure; such a configuration might be relevant to star formation if prestellar cloud cores when formed are nearly in equilibrium and in approximate pressure balance with a surrounding medium. For a fixed sound speed *c* and boundary pressure *P*, an isothermal sphere is unstable to collapse if its radius and mass exceed the critical values

$$R_{\rm BE} = \frac{0.48 \, c^2}{G^{1/2} P^{1/2}}, \ M_{\rm BE} = \frac{1.18 \, c^4}{G^{3/2} P^{1/2}}$$

(Bonnor-Ebert sphere).

These results can be related to the Jeans length and mass discussed above by noting that in an isothermal medium the pressure and density are related by  $P = \rho c^2$  (*c* is sound speed);

thus,  $R_{\rm BE}$  and  $M_{\rm BE}$  have the same dimensional form as the Jeans length and mass, but with smaller numerical coefficients that reflect the fact that a Bonnor–Ebert sphere contains only matter whose density is higher than the background density, while a region one Jeans length across also includes matter of lower density that may or may not collapse along with the denser material.

(R.B. Larson, The physics of star formation, Rep. Prog. Phys. 66, 1651-1697, 2003)









