

The non-LTE Rate Equations

Statistical equations

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Population numbers

LTE: population numbers follow from Saha-Boltzmann equations, i.e. purely **local** problem

$$n_i^* = n_i^*(T, n_e)$$

Non-LTE: population numbers also depend on radiation field. This, in turn, is depending on the population numbers in all depths, i.e. **non-local** problem.

$$n_i = n_i(T, n_e, J)$$

The Saha-Boltzmann equations are replaced by a detailed consideration of atomic processes which are responsible for the population and de-population of atomic energy levels:

Excitation and de-excitation
Ionization and recombination

} by radiation or collisions

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Statistical Equilibrium

Change of population number of a level with time:

= Sum of all **population processes** into this level

- Sum of all **de-population processes** out from this level

$$\frac{d}{dt} n_i = \sum_{j \neq i} n_j P_{ji} - n_i \sum_{j \neq i} P_{ij}$$

$$\begin{aligned} \frac{d}{dt} n_i &= \sum_{j \neq i} n_j P_{ji} \\ &- n_i \sum_{j \neq i} P_{ij} \end{aligned}$$

One such equation for each level

The transition rate P_{ij} comprises radiative rates R_{ij}
and collision rates C_{ij}

In stellar atmospheres we often have the stationary case:

$$\frac{d}{dt} n_i = 0 \quad \text{hence} \quad \boxed{\sum_{j \neq i} n_j P_{ji} = n_i \sum_{j \neq i} P_{ij}} \quad \text{for all levels } i$$

These equations determine the population numbers.

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Radiative rates: bound-bound transitions

Two alternative formulations:

a) **Einstein coefficients** $B_{ij} \ B_{ji} \ A_{ji}$

b) **Line absorption coefficients** $\sigma_{ij}(\nu)$

advantage a): useful for analytical expressions with simplified model atoms

advantage b): similar expressions in case of bound-free transitions: good for efficient programming

Number of transitions $i \rightarrow j$ induced by intensity I_ν in frequency interval $d\nu$ and solid angle $d\omega$

$$n_i B_{ij} \phi_\nu I_\nu d\nu d\omega / 4\pi \quad (\text{absorbed Energy} / h\nu)$$

Integration over frequencies and angles yields

$$n_i R_{ij} = n_i B_{ij} \int_0^\infty \phi_\nu J_\nu d\nu$$

Or alternatively

$$\text{with } \sigma_{ij}(\nu) = B_{ij} \phi(\nu) h\nu / 4\pi \quad \boxed{n_i R_{ij} = n_i 4\pi \int_0^\infty \frac{\sigma_{ij}(\nu)}{h\nu} J_\nu d\nu}$$

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Radiative rates: bound-bound transitions

In analogy, number of stimulated emissions:

$$n_j R'_{ji} = n_j B_{ji} \int_0^{\infty} \varphi_\nu J_\nu d\nu = n_j B_{ij} \frac{g_i}{g_j} \int_0^{\infty} \varphi_\nu J_\nu d\nu$$

$$n_j R'_{ji} = n_j 4\pi \frac{g_i}{g_j} \int_0^{\infty} \frac{\sigma_{ij}}{h\nu} J_\nu d\nu$$

Number of spontaneous emissions:

$$n_j R''_{ji} = n_j \int_0^{\infty} A_{ji} \varphi_\nu d\nu = n_j B_{ji} \int_0^{\infty} \frac{2h\nu^3}{c^2} \varphi_\nu d\nu$$

$$n_j R''_{ji} = n_j 4\pi \frac{g_i}{g_j} \int_0^{\infty} \frac{\sigma_{ij}}{h\nu} \frac{2h\nu^3}{c^2} d\nu$$

Total downwards rate:

$$n_j \bar{R}_{ji} = n_j (R'_{ji} + R''_{ji}) = n_j 4\pi \frac{g_i}{g_j} \int_0^{\infty} \frac{\sigma_{ij}}{h\nu} \left(\frac{2h\nu^3}{c^2} + J_\nu \right) d\nu$$

$$n_j \bar{R}_{ji} = n_j \left(\frac{n_i}{n_j} \right)^* R_{ji} = n_j \left(\frac{n_i}{n_j} \right)^* [4\pi] \int_0^{\infty} \frac{\sigma_{ij}}{h\nu} \left(\frac{2h\nu^3}{c^2} + J_\nu \right) e^{-h\nu/kT} d\nu$$

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Radiative rates: bound-free transitions

Also possible: ionization into excited states of parent ion

Example C III:

Ground state $2s^2 \ ^1S$

Photoionisation produces C IV in ground state $2s \ ^2S$

C III in first excited state $2s2p \ ^3P^o$

Two possibilities:

Ionization of 2p electron \rightarrow C IV in ground state $2s \ ^2S$

Ionization of 2s electron \rightarrow C IV in first excited state $2p \ ^2P$

C III two excited electrons, e.g. $2p^2 \ ^3P$

Photoionization only into excited C IV ion $2p \ ^2P$

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Radiative rates: bound-free transitions

Number of photoionizations = absorbed energy in $d\nu$, divided by photon energy, integrated over frequencies and solid angle

$$\int_0^\infty \oint n_i p_\nu I_\nu d\omega d\nu \rightarrow n_i R_{ij} = n_i 4\pi \int_0^\infty \frac{\sigma_{ij}(\nu)}{h\nu} J_\nu d\nu$$



Number of spontaneous recombinations:

$$\int_0^\infty \oint n_j n_e(\nu) F(\nu) d\omega d\nu \rightarrow n_j R_{ji} = n_j 4\pi \int_0^\infty n_e(\nu) \frac{2h\nu^3}{c^2} G(\nu) \frac{h}{m} d\nu$$



$$n_j R_{ji} = n_j 4\pi \int_0^\infty n_e(\nu) \frac{2h\nu^3}{c^2} p_\nu \frac{m}{h} e^{-h\nu/kT} \left(\frac{n_i}{n_j}\right)^* \frac{1}{n_e(\nu)} \frac{h}{m} d\nu$$

$$n_j R_{ji} = n_j \left(\frac{n_i}{n_j}\right)^* 4\pi \int_0^\infty \frac{\sigma_{ij}(\nu)}{h\nu} \frac{2h\nu^3}{c^2} e^{-h\nu/kT} d\nu$$

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Radiative rates: bound-free transitions

Number of induced recombinations

$$\int_0^\infty \oint n_j n_e(\nu) G(\nu) I_\nu d\omega d\nu \rightarrow n_j R_{ji} = n_j 4\pi \int_0^\infty n_e(\nu) G(\nu) J_\nu \frac{h}{m} d\nu$$

$$n_j R_{ji} = n_j 4\pi \int_0^\infty n_e(\nu) p_\nu \frac{m}{h} e^{-h\nu/kT} \left(\frac{n_i}{n_j}\right)^* \frac{1}{n_e(\nu)} J_\nu \frac{h}{m} d\nu$$

$$n_j R_{ji} = n_j \left(\frac{n_i}{n_j}\right)^* 4\pi \int_0^\infty \frac{\sigma_{ij}(\nu)}{h\nu} J_\nu e^{-h\nu/kT} d\nu$$

Total recombination rate

$$n_j \left(\frac{n_i}{n_j}\right)^* R_{ji} = n_j \left(\frac{n_i}{n_j}\right)^* 4\pi \int_0^\infty \frac{\sigma_{ij}(\nu)}{h\nu} \left(\frac{2h\nu^3}{c^2} + J_\nu\right) e^{-h\nu/kT} d\nu$$

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Radiative rates

Upward rates:

$n_i R_{ij}$ with

$$R_{ij} = 4\pi \int_0^{\infty} \frac{\sigma_{ij}(\nu)}{h\nu} J_{\nu} d\nu$$

Downward rates:

$n_j \left(\frac{n_i}{n_j} \right)^* R_{ji}$ with

$$R_{ji} = 4\pi \int_0^{\infty} \frac{\sigma_{ij}(\nu)}{h\nu} \left(\frac{2h\nu^3}{c^2} + J_{\nu} \right) e^{-h\nu/kT} d\nu$$

Remark: in TE we have $J_{\nu} = B_{\nu} \Rightarrow R_{ij}^* = R_{ji}^* \Rightarrow \frac{n_i}{n_j} = \left(\frac{n_i}{n_j} \right)^*$

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Collisional rates

Stellar atmosphere: Plasma, with atoms, ions, electrons

Particle collisions induce excitation and ionization

Cool stars: matter mostly neutral \Rightarrow frequent collisions with neutral hydrogen atoms

Hot stars: matter mostly ionized \Rightarrow collisions with ions become important; but much more important become **electron collisions**

$$\frac{v_{electron}}{v_{ion}} = \left(\frac{\text{ion mass}}{\text{electron mass}} \right)^{1/2} = \left(\frac{m_H A}{m_e} \right)^{1/2} \approx 43\sqrt{A}$$

Therefore, in the following, we only consider collisions of atoms and ions with electrons.

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Electron collisional rates

Transition $i \rightarrow j$ (j : bound or free), $\sigma_{ij}(v)$ = electron collision cross-section, v = electron speed

Total number of transitions $i \rightarrow j$:

$$n_i C_{ij} = n_i n_e \int_{v_0}^{\infty} \sigma_{ij}(v) f(v) v dv = n_i n_e \Omega_{ij}(T)$$

v_0 minimum velocity necessary for excitation (threshold)
 $f(v)dv$ velocity distribution (Maxwell)

In TE we have therefore

$$n_i^* C_{ij} = n_j^* C_{ji}$$

Total number of transitions $j \rightarrow i$:

$$n_j C_{ji} = n_j \left(\frac{n_i}{n_j} \right)^* C_{ij}$$

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Electron collisional rates

We look for: collisional cross-sections $\sigma_{ij}(v)$

- experiments
- quantum mechanical calculations

Usually: Bohr radius πa_0^2 as unit for cross-section $\sigma_{ij}(v)$

$$\sigma_{ij}(v) = \pi a_0^2 Q_{ij}$$

Q_{ij} usually tabulated as function of energy of colliding electron

$$\Omega_{ij}(T) = \int_{v_0}^{\infty} \sigma_{ij}(v) f(v) v dv \quad \text{with } 1/2 m v^2 = E \quad \text{and } f(v) dv = \left(\frac{m_e}{2\pi kT} \right)^{3/2} e^{-mv^2/2kT} 4\pi v^2 dv$$

$$= C_0 \sqrt{T} \int_{u_0}^{\infty} Q_{ij}(ukT) u e^{-u} du \quad \text{with } u = E/kT \quad \text{and } C_0 = \pi a_0^2 \left(\frac{8k}{m_e \pi} \right)^{1/2} = 5.456 \cdot 10^{-11}$$

$$\Omega_{ij}(T) = C_0 \sqrt{T} e^{-u_0} \Gamma_{ij}(T) \quad \text{with } \Gamma_{ij}(T) = \int_0^{\infty} Q_{ij}(E_0 + xkT) (x + u_0) e^{-x} dx, \quad x := (E/kT - E_0/kT)$$

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Electron collisional rates

$$\Omega_{ij}(T) = C_0 \sqrt{T} e^{-u_0} \Gamma_{ij}(T)$$

$$n_i C_{ij} = n_i n_e \Omega_{ij}(T)$$

Advantage of this choice of notation:

Main temperature dependence is described by $\sqrt{T} e^{-E_0/kT}$

$\Gamma_{ij}(T)$ only weakly varying function of T

Hence, simple polynomial fit possible

⇒ Important for numerical application

Now: examples how to compute the C_{ij}

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Computation of collisional rates: Excitation

Van Regemorter (1962): Very useful approximate formula for allowed dipole transitions

$$C_{ij} = C_0 n_e \sqrt{T} 14.5 f_{ij} \left(\frac{E_H}{E_0} \right)^2 u_0 e^{-u_0} \Gamma(u_0)$$

E_H hydrogen ionization energy

f_{ij} oscillator strength of radiative transition

$$E_0 = h\nu_{ij} \quad u_0 = E_0 / kT \quad \Gamma(u_0) = \max[\bar{g}, 0.276 e^{u_0} E_1(u_0)]$$

$$\bar{g} = \begin{cases} 0.7 & \text{for transitions between levels with equal principal quantum number} \\ 0.2 & \text{else} \end{cases}$$

There exist many formulae, made for particular ions and transitions, e.g., (optically) forbidden transitions between n=2 levels in He I (Mihalas & Stone 1968)

$$C_{ij} = C_0 n_e \sqrt{T} e^{-u_0} \Gamma(T)$$

with $\log \Gamma = c_0 + c_1 \log T + c_{-2} (\log T)^{-2}$ coefficients **c** tabulated for each transition

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Computation of collisional rates: Ionization

The **Seaton formula** is in analogy to the van-Regemorter formula in case of excitation. Here, the photon absorption cross-section for ionization is utilized:

$$C_{ij} = 1.55 \cdot 10^{13} \sigma_0 \bar{g} \frac{n_e}{\sqrt{T}} \frac{e^{-u_0}}{u_0}$$

σ_0 = threshold photon cross-section for ionization

$$\bar{g} = \begin{cases} 0.1 & \text{for ions with charge } Z = 1 \\ 0.2 & \text{for ions with charge } Z = 2 \\ 0.3 & \text{for ions with charge } Z > 2 \end{cases}$$

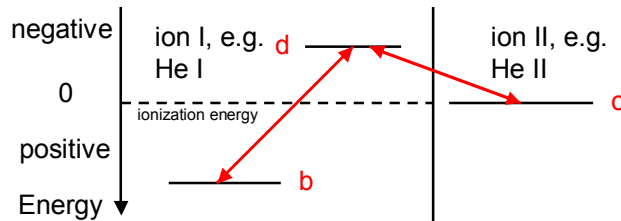
Alternative: semi-empirical formula by Lotz (1968):

$$C_{ij} = C_0 n_e \sqrt{T} 2.5 a \left(\frac{E_H}{E_0} \right)^2 u_0 [E_1(u_0) - b e^c u_0 E_1(u_1) / u_1]$$

$u_1 = u_0 + c$ a, b, c empirical quantities, adjusted to individual atoms

For H and He specific fit formulae are used, mostly from Mihalas (1967) and Mihalas & Stone (1968)

Autoionization and dielectronic recombination



b bound state, **d** doubly excited state, autoionization level

c ground state of next Ion

d → **c**: **Autoionization**. **d** decays into ground state of next ionization stage plus free electron

c → **d** → **b**: **Dielectronic recombination**. Recombination via a doubly excited state of next lower ionization stage. **d** auto-ionizes again with high probability: $A_{\text{auto}} = 10^{13} \dots 10^{14} / \text{sec}$! But sometimes a stabilizing transition **d** → **b** occurs, by which the excited level decays radiatively.

Computation of rates

Number of dielectronic recombinations from **c** to **b**:

$$n_c R_{cb} = n_d A_s \quad A_s = \text{probability for spontaneous stabilizing transition}$$

In the limit of weak radiation fields the reverse process can be neglected. Then we obtain (Bates 1962):

$$n_d = n_d^* A_a / (A_a + A_s) \quad \text{with } n_d^* = n_c n_e C_1 T^{-3/2} e^{E_{\text{ion}}^d / kT} = n_c n_e \Phi_{cd}(T)$$

A_a = transition probability for autoionization

So, the number of dielectronic recombinations from **c** to **b** is:

$$n_c R_{cb} = n_c n_e \Phi_{cd}(T) A_s A_a / (A_a + A_s)$$

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Computation of rates

There are two different regimes:

a) high temperature dielectronic recombination **HTDR**

b) low temperature dielectronic recombination **LTDR**

for the cases that the autoionizing levels are close to the ionization limit (**b**) or far above it (**a**)

a) Most important recombination process He II \rightarrow He I in the solar corona ($T \sim 2 \cdot 10^6 \text{K}$)

b) Very important for specific ions in photospheres ($T < 10^5 \text{K}$)
e.g. N III $\lambda 4634\text{-}40\text{\AA}$ emission complex in Of stars

Reason: upper level is overpopulated, because a stabilizing transition is going into it.

Because in case **b)** $A_a \gg A_s \rightarrow n_d = n_d^*$

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LTDR

The radiation field in photospheres is **not** weak, i.e., the reverse process **b** → **d** is induced

Recombination rate:

$$n_c R_{cb} = n_c n_e \Phi_{cd}(T) A_s \left(1 + \frac{c^2}{2h\nu^3} \bar{J} \right)$$

\bar{J} mean intensity in stabilizing transition, i.e.,

given by continuum value (line very broad, because short lifetime)

Reverse process:

$$n_b R_{bc} = n_b B_{bd} \bar{J} = n_b A_s \frac{c^2}{2h\nu^3} \frac{g_d}{g_b} \bar{J}$$

These rates are formally added to the usual ionization and recombination rates and do not show up explicitly in the rate equations.

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Complete rate equations

For each atomic level **i** of each ion, of each chemical element we have:

$$n_i \sum_{j \neq i} P_{ij} - \sum_{j \neq i} n_j P_{ji} = 0$$

In detail:

$$\begin{aligned}
 & n_i \left[\sum_{j>i} (R_{ij} + C_{ij}) \right. \\
 & \quad \left. + \sum_{j<i} \left(\frac{n_j}{n_i} \right)^* (R_{ij} + C_{ji}) \right] \\
 & - \sum_{j>i} n_j \left(\frac{n_i}{n_j} \right)^* (R_{ji} + C_{ij}) \\
 & - \sum_{j<i} n_j (R_{ji} + C_{ji}) \\
 & = 0
 \end{aligned}$$

↑ excitation and ionization

↓ de-excitation and recombination

↓ de-excitation and recombination

↑ excitation and ionization

rates out of *i*

rates into *i*

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Closure equation

One equation for each chemical element is redundant, e.g., the equation for the highest level of the highest ionization stage; to see this, add up all equations except for the final one: these rate equations only yield population **ratios**.

We therefore need a **closure equation** for each chemical species:

Abundance definition equation of species k , written for example as number abundance y_k relative to hydrogen:

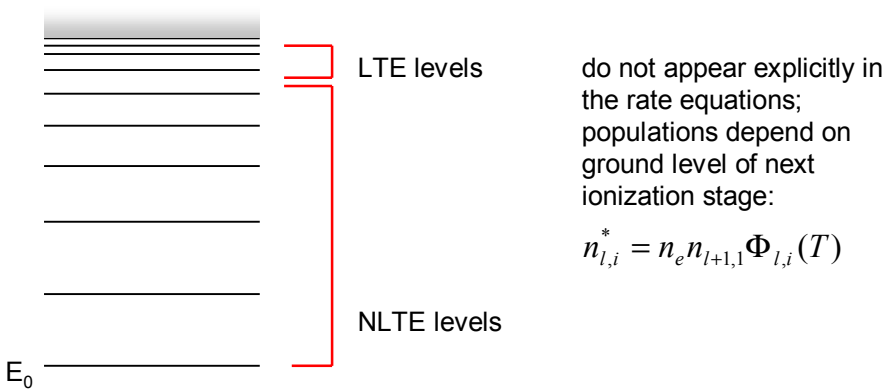
$$y_k = \frac{\sum \text{population numbers of species } k}{\sum \text{population numbers of hydrogen}}$$

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Abundance definition equation

Notation:

Population number of level i in ionization stage l : $n_{l,i}$



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Abundance definition equation

Notation:

$NION$ number of ionization stages of chemical element k

$NL(l)$ number of NLTE levels of ion l

$LTE(l)$ number of LTE levels of ion l

$$\sum_{l=1}^{NION} \left[\sum_{i=1}^{NL(l)} n_{l,i} + \sum_{i=1}^{LTE(l)} n_{l,i}^* \right] = y_k \left[\sum_{i=1}^{NL(H)} n_i + \sum_{i=1}^{LTE(H)} n_i^* + n_{protons} \right] \Rightarrow$$

$$\sum_{l=1}^{NION} \left[\sum_{i=1}^{NL(l)} n_{l,i} + n_{l+1,1} n_e \sum_{i=1}^{LTE(l)} \Phi_{li}(T) \right] = y_k \left[\sum_{i=1}^{NL(H)} n_{l,i} + n_{protons} \left(1 + n_e \sum_{i=1}^{LTE(H)} \Phi_i(T) \right) \right]$$

Also, one of the abundance definition equations is **redundant**, since abundances are given relative to hydrogen (other definitions don't help) \Rightarrow **charge conservation**

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Charge conservation equation

Notation:

Population number of level i , ion l , element k : n_{kli}

$NELEM$ number of chemical elements

$q(l)$ charge of ion l

$$\sum_{k=1}^{NELEM} \sum_{l=1}^{NION} q(l) \left[\sum_{i=1}^{NL(l,k)} n_{kli} + \sum_{i=1}^{LTE(l,k)} n_{kli}^* \right] = n_e \Rightarrow$$

$$\sum_{k=1}^{NELEM} \sum_{l=1}^{NION} q(l) \left[\sum_{i=1}^{NL(l,k)} n_{kli} + n_{k,l+1,1} n_e \sum_{i=1}^{LTE(l)} \Phi_{kli}(T) \right] = n_e$$

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Complete rate equations: Matrix notation

Vector of population numbers

$$\underline{n} = (n_1, n_2, \dots, n_{\text{NLALL}}) \quad \text{NLALL} = \text{total number of NLTE levels}$$

$$\underline{A} \underline{n} = \underline{b} \quad \text{rate equation in matrix notation}$$

One such system of equations per depth point

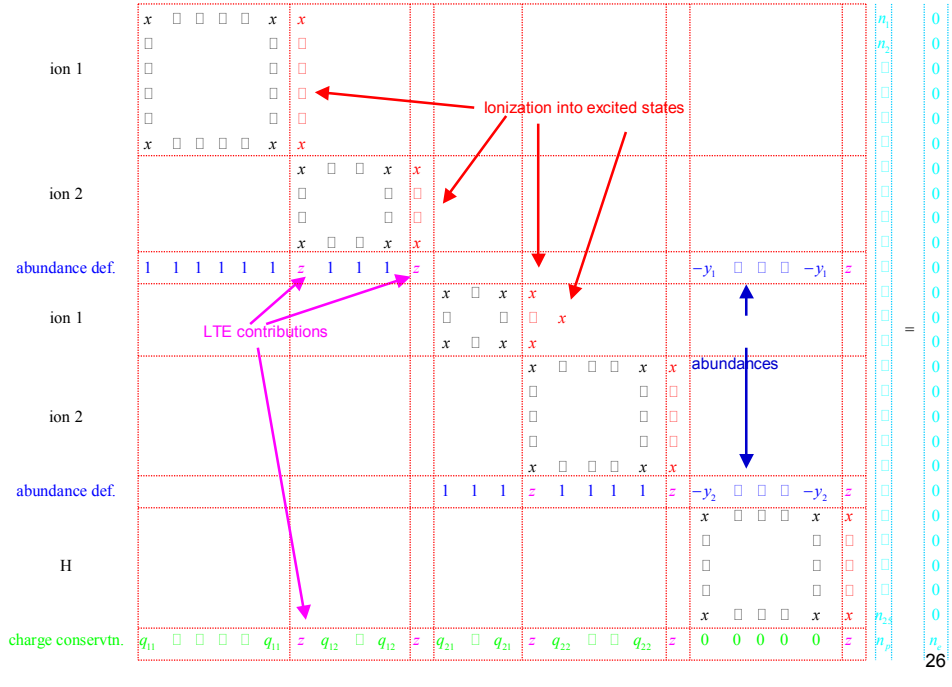
Example: 3 chemical elements

Element 1: NLTE-levels: ion1: 6, ion2: 4, ion3: 1

Element 2: NLTE-levels: ion1: 3, ion2: 5, ion3: 1

Element 3: NLTE-levels: ion1: 5, ion2: 1, hydrogen

Number of levels: NLALL=26, i.e. 26 x 26 matrix



Elements of rate matrix

For each ion l with $NL(l)$ NLTE levels one obtains a sub-matrix with the following elements:

$$A_{ij} = \begin{cases} -(R_{ji} + C_{ji}) & j < i \text{ lower left} \\ -\left(\frac{n_i}{n_j}\right)^* (R_{ji} + C_{ij}) & j > i \text{ upper right} \\ \sum_{m < i} \left(\frac{n_m}{n_i}\right)^* (R_{im} + C_{mi}) + \sum_{m > i}^k (R_{im} + C_{im}) & j = i \text{ diagonal} \end{cases}$$

$i = 1 \dots NL(l)$ $j = 1 \dots NL(l) \dots k$ k highest level in parent ion, into which ion l can ionize; does not have to be $= NL(l) + 1$!

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Elements corresponding to abundance definition eq.

Are located in final row of the respective element:

$$i = \sum_{l=1}^{MON} NL(l)$$

$$A_{ij} = \begin{cases} 1 & j = 1 \dots \sum_{l=1}^{MON} NL(l) \text{ except of ground state of excited ions} \\ 1 + n_e \sum_{m=LTE(l-1)} \Phi_{l-1,m} & j = \text{ground state of excited ions} \\ -y_k & j = [NLALL - NL(H)] \dots [NLALL - 1] \\ -y_k \left(1 + n_e \sum_{m=LTE(H)} \Phi_m \right) & j = NLALL \end{cases}$$

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Elements corresponding to charge conservation eq.

Are located in the very final row of rate matrix, i.e., in

$$i = NLALL$$

$$A_{ij} = \begin{cases} q(l) & j = 1 \dots NLALL, \text{ except of} \\ & \text{ground state of excited ions} \\ q(l) + q(l-1) \sum_{m=LTE(-1)} \Phi_{l-1,m} & \text{else} \end{cases}$$

Note: the inhomogeneity vector **b** (right-hand side of statistical equations) contains zeros except for the very last element (i=NLALL): electron density n_e (from charge conservation equation)

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Solution by linearization

The equation system $\underline{\underline{A}}\underline{n} = \underline{b}$ is a linear system for \underline{n} and can be solved if, n_e, T, J_ν are known. But: these quantities are in general **unknown**. Usually, only approximate solutions within an **iterative** process are known.

Let all these variables change by $\delta n_e, \delta T, \delta J_\nu$ e.g. in order to fulfill energy conservation or hydrostatic equilibrium.

Response of populations $\delta \underline{n}$ on such changes:

Let $\underline{\underline{A}}\underline{n} = \underline{b}$ with actual quantities

And $(\underline{\underline{A}} + \delta \underline{\underline{A}})(\underline{n} + \delta \underline{n}) = (\underline{b} + \delta \underline{b})$ with new quantities n_e, T, J_ν

Neglecting 2nd order terms, we have:

$$\underline{\underline{A}}\underline{n} - \underline{b} = -\delta \underline{n} - \underline{n} \delta \underline{\underline{A}} + \delta \underline{b}$$

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Linearization of rate equations

Needed: expressions for: $\underline{\delta A}, \underline{\delta b}$ Jv discretized in NF frequency points

One possibility:

$$\underline{\delta A} = \frac{\partial \underline{A}}{\partial T} \delta T + \frac{\partial \underline{A}}{\partial n_e} \delta n_e + \sum_{k=1}^{NF} \frac{\partial \underline{A}}{\partial J_k} \delta J_k$$

If in addition to \underline{n} the variables n_e, T, J_k are introduced as unknowns, then we have the

→ Method of Complete Linearization

Other possibility: eliminates J_k from the equation system by expressing J_k through the other variables n_e, T :

$$J_k = f(\underline{n}, T, n_e)$$

As an approximation one uses $J_k^d \sim S_k^d(\underline{n}, T, n_e)$
(and iterates for exact solution)

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Linearization of rate equations

$$\underline{\delta A} = \frac{\partial \underline{A}}{\partial T} \delta T + \frac{\partial \underline{A}}{\partial n_e} \delta n_e + \sum_{k=1}^{NF} \frac{\partial \underline{A}}{\partial S_k} \delta S_k$$

$$\underline{\delta S}_k = \frac{\partial S_k}{\partial T} \delta T + \frac{\partial S_k}{\partial n_e} \delta n_e + \sum_{j=1}^{NLALL} \frac{\partial S_k}{\partial n_j} \delta n_j$$

Method of approximate Λ -operators (Accelerated Lambda Iteration)

analogous, $\underline{\delta b}$:

$$\underline{\delta b} = \frac{\partial \underline{b}}{\partial T} \delta T + \frac{\partial \underline{b}}{\partial n_e} \delta n_e + \sum_{k=1}^{NF} \frac{\partial \underline{b}}{\partial S_k} \delta S_k = (0, \dots, 0, \delta n_e)$$

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Linearization of rate equations

$$\begin{aligned} \underline{\underline{A}}n - \underline{\underline{b}} = -\delta n \underline{\underline{A}} & + \delta T \left[-\frac{\partial \underline{\underline{A}}}{\partial T} n - \sum_{k=1}^{NF} n \frac{\partial \underline{\underline{A}}}{\partial S_k} \frac{\partial S_k}{\partial T} \right] \\ & + \delta n_e \left[-\frac{\partial \underline{\underline{A}}}{\partial n_e} n - \sum_{k=1}^{NF} n \frac{\partial \underline{\underline{A}}}{\partial S_k} \frac{\partial S_k}{\partial n_e} \right] \\ & + \sum_{j=1}^{NLALL} \delta n_j \left[-\sum_{k=1}^{NF} n \frac{\partial \underline{\underline{A}}}{\partial S_k} \frac{\partial S_k}{\partial n_j} \right] \end{aligned}$$

Linearized equation for response δn as answer on changes $\delta n_e, \delta T, \delta J_\nu$

Expressions $\sum_{k=1}^{NF} n \frac{\partial \underline{\underline{A}}}{\partial S_k} \frac{\partial S_k}{\partial n_j}$ show the complex coupling of all variables. A change in the radiation field and, hence, the source function **at any frequency** causes a change of populations of **all levels**, even if a particular level cannot absorb or emit a photon at that very frequency!

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Linearization of rate equations

In order to solve the linearized rate equations we need to compute these derivatives:

$$\frac{\partial}{\partial n_e}, \frac{\partial}{\partial T}, \frac{\partial}{\partial n_j}, \frac{\partial}{\partial S_k} \quad \text{with respect to } \underline{\underline{A}}, \underline{\underline{b}}, S_k$$

All derivatives can be computed **analytically!**

Increases accuracy and stability of numerical solution. More details later.

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LTE or NLTE?

When do departures from LTE become important?

LTE is a good approximation, if:

- 1) Collisional rates dominate for all transitions

$$R_{ij} \ll C_{ij} \rightarrow P_{ij} = R_{ij} + C_{ij} \approx C_{ij}$$

$$\text{because } \frac{C_{ij}}{C_{ji}} = \left(\frac{n_i}{n_j} \right)^*$$

solution of rate equations \rightarrow LTE

- 2) $J_\nu = B_\nu$ is a good approximation at all frequencies

$$R_{ij} = R_{ji}$$

$$\rightarrow \frac{n_i}{n_j} = \left(\frac{n_i}{n_j} \right)^*$$

solution of rate equations \rightarrow LTE

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LTE or NLTE?

When do departures from LTE become important?

LTE is a bad approximation, if:

- 1) Collisional rates are small $C_{ij} \sim n_e / \sqrt{T}$ $n_e \downarrow, T \uparrow \Rightarrow C_{ij} \downarrow$
- 2) Radiative rates are large $R_{ij} \sim T^\alpha, \alpha > 1$ $T \uparrow \Rightarrow R_{ij} \uparrow$
- 3) Mean free path of photons is larger than that of electrons

Example: pure hydrogen plasma

$$\Delta z \sim 1/n_H \text{ (density of neutral H)}$$

$$\text{Saha: } n_H \sim n_e n_p T^{-3/2} e^{\Delta E/kT} \rightarrow \Delta z \sim \frac{T^{3/2}}{n_e n_p} e^{-\Delta E/kT}$$

$$n_e \downarrow, T \uparrow \Rightarrow \Delta z \uparrow$$

Departures from LTE occur, if temperatures are high and densities are low

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LTE or NLTE?

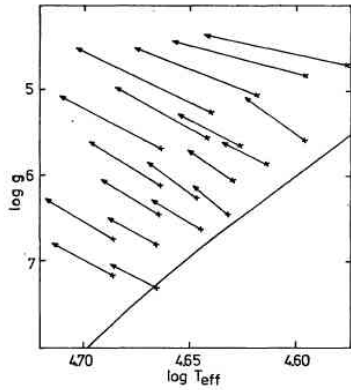


Fig. 8. "Non-LTE vectors" [Displacement due to non-LTE effects in the $(\log g, \log T_{\text{eff}})$ -diagram] for $N(\text{He})/N(\text{H})=0.1$

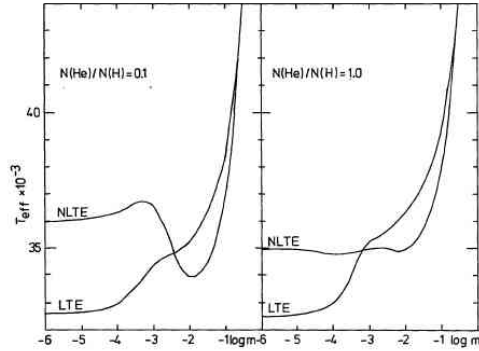


Fig. 4. Temperature stratification in NLTE and LTE for $T_{\text{eff}}=45000$ K, $\log g=5$ and two different helium abundances ($N(\text{He})/N(\text{H})=0.1$ and 1.0)

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LTE or NLTE?

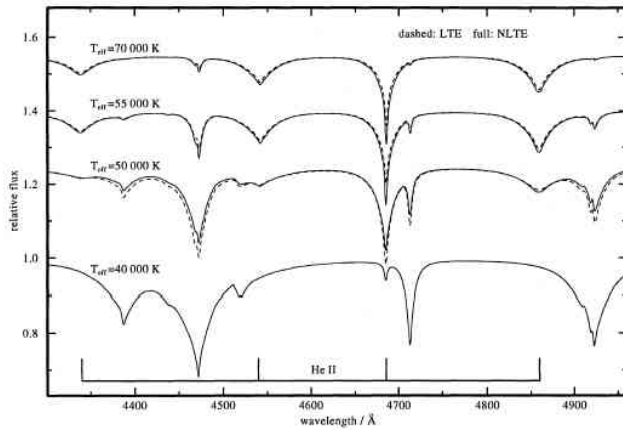
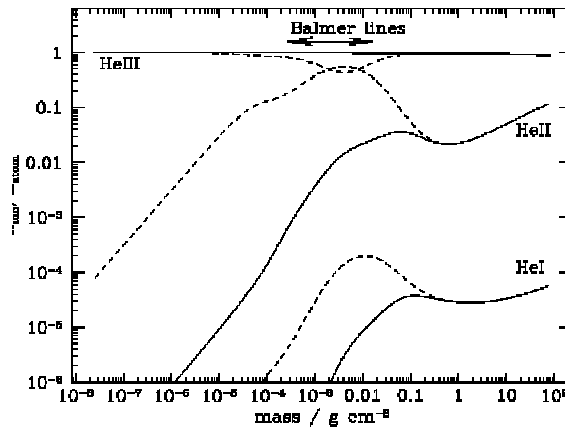
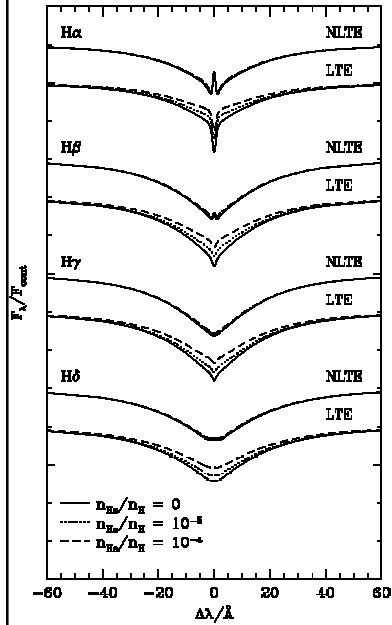


Fig. 3. NLTE effects on synthetic DO line profiles (convoluted by a 2 \AA Gauss profile) at various effective temperatures. The two hotter models are calculated at $\log g=7.5$, the cooler ones at $\log g=8.0$. Significant deviations between LTE (dashed) and NLTE spectra occur down to $T_{\text{eff}}=50000$ K. At $T_{\text{eff}}=40000$ K the NLTE effects disappear completely, justifying LTE analyses for DB white dwarfs

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LTE or NLTE?

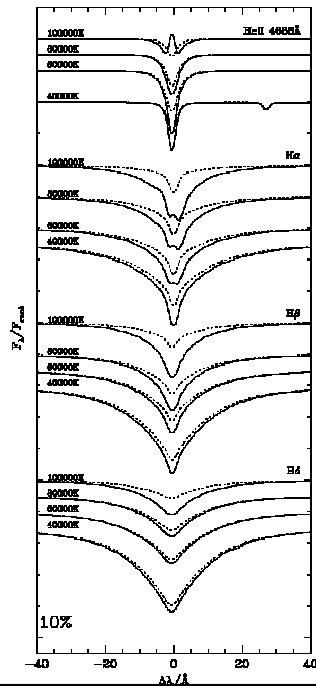
DA white dwarf, $T_{\text{eff}} = 60000\text{K}$, $\log g = 7.5$



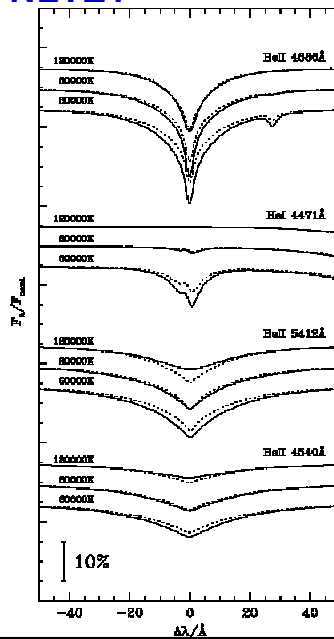
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LTE or NLTE?

DAO with $\log g = 6.5$



DO with $\log g = 7.5$



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Summary: non-LTE Rate Equations

Complete rate equations

For each atomic level i of each ion, of each chemical element we have:

$$n_i \sum_{j \neq i} P_{ij} - \sum_{j \neq i} n_j P_{ji} = 0$$

In detail:

$$\begin{aligned}
 & n_i \left[\sum_{j>i} (R_{ij} + C_{ij}) \right. \\
 & \quad \left. + \sum_{j<i} \left(\frac{n_j}{n_i} \right)^* (R_{ij} + C_{ji}) \right] \begin{array}{l} \uparrow \text{excitation and ionization} \\ \text{rates out of } i \\ \downarrow \text{de-excitation and recombination} \end{array} \\
 & - \sum_{j>i} n_j \left(\frac{n_i}{n_j} \right)^* (R_{ji} + C_{ij}) \begin{array}{l} \downarrow \text{de-excitation and recombination} \\ \text{rates into } i \end{array} \\
 & - \sum_{j<i} n_j (R_{ji} + C_{ji}) \begin{array}{l} \uparrow \text{excitation and ionization} \end{array} \\
 & = 0
 \end{aligned}$$